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PPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION N
09/954,656	09/18/2001	Georg Friedrich Gaertner	DE000145	7505
24737	590 06/29/2004		EXAMINER	
	TELLECTUAL PROPE	SANTIAGO, MARICELI		
P.O. BOX 3001 BRIARCLIFF MANOR, NY 10510			ART UNIT	PAPER NUMBER
	ŕ		2879	
		DATE MAILED: 06/29/2004		

Please find below and/or attached an Office communication concerning this application or proceeding.

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	Application No.	Applicant(s)
	09/954,656	GAERTNER ET AL.
Office Action Summary	Examiner	Art Unit
	Mariceli Santiago	2879
The MAILING DATE of this communication app Period for Reply	pears on the cover sheet with the c	rrespondence address
A SHORTENED STATUTORY PERIOD FOR REPL' THE MAILING DATE OF THIS COMMUNICATION. - Extensions of time may be available under the provisions of 37 CFR 1.1 after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a repl - If NO period for reply is specified above, the maximum statutory period - Failure to reply within the set or extended period for reply will, by statute Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	36(a). In no event, however, may a reply be time y within the statutory minimum of thirty (30) days will apply and will expire SIX (6) MONTHS from the cause the application to become ABANDONE	nely filed s will be considered timely. the mailing date of this communication. D (35 U.S.C. § 133).
Status		
 Responsive to communication(s) filed on <u>15 A</u> This action is FINAL. 2b) ☐ This Since this application is in condition for allowarclosed in accordance with the practice under B 	s action is non-final. nce except for formal matters, pro	
Disposition of Claims		
 4) Claim(s) 1-9 is/are pending in the application. 4a) Of the above claim(s) is/are withdram 5) Claim(s) 5,6 and 8 is/are allowed. 6) Claim(s) 1-4,7 and 9 is/are rejected. 7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and/or 		
Application Papers		
9) The specification is objected to by the Examine 10) The drawing(s) filed on 18 September 2001 is/s Applicant may not request that any objection to the Replacement drawing sheet(s) including the correct 11) The oath or declaration is objected to by the Example 11.	are: a) \square accepted or b) \square objection drawing(s) be held in abeyance. See tion is required if the drawing(s) is objection is required.	e 37 CFR 1.85(a). lected to. See 37 CFR 1.121(d).
Priority under 35 U.S.C. § 119		
 12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of: 1. Certified copies of the priority document 2. Certified copies of the priority document 3. Copies of the certified copies of the priority application from the International Burea * See the attached detailed Office action for a list 	is have been received. Is have been received in Application of the second in the secon	on No ed in this National Stage
Attachment(s) 1) D Notice of References Cited (PTO-892)	4) Interview Summary	
 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) Paper No(s)/Mail Date 	Paper No(s)/Mail Da	

DETAILED ACTION

Response to Amendment

The Amendment, filed on April 15, 2004, has been entered and acknowledged by the Examiner.

Claims 1-9 are pending in the instant application.

Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 1, 2 and 9 are rejected under 35 U.S.C. 102(b) as being anticipated by Derks et al. (US 5,075,589).

Regarding claim 1, Derks discloses a cathode ray tube provided with at least one oxide cathode (Fig. 1) comprising a cathode carrier with a cathode base (7) of a cathode metal and a cathode coating of an electron-emitting material (2) containing a particle-particle composite material of oxide particles of an alkaline earth oxide selected from the group formed by the oxides of calcium, strontium and barium (Column 2, lines 49-52), and oxide particles having a first grain size distribution of an oxide selected from the group formed by the oxides of scandium, yttrium and the lanthanoids (Column 2, lines 27-29, $d_{50} > 0.9 \mu m$), and oxide particles having a second grain size distribution of an oxide selected from the group formed by the oxides of scandium, yttrium and the lanthanoids (Column 2, lines 27-29, $d_{50} \le 0.9 \mu m$).

Regarding claim 2, Derks discloses a cathode ray tube wherein the oxide particles having the first grain size distribution have and average grain size $d_{50} > 0.9 \mu m$ (Column 2, lines

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27-29, i.e., the value falls within the claimed size distribution of $0.4 < d_{50} < 5\mu m$), and the oxide particles having the second grain size distribution have an average grain size $d_{50} \le 0.9\mu m$ (Column 2, lines 27-29, i.e., the value falls within the claimed size distribution of $d_{50} < 0.4\mu m$),

Regarding claim 9, Derks discloses an oxide cathode (Fig. 1) comprising a cathode carrier with a cathode base (7) of a cathode metal and a cathode coating of an electron-emitting material (2) containing a particle-particle composite material of oxide particles of an alkaline earth oxide selected from the group formed by the oxides of calcium, strontium and barium (Column 2, lines 12-19), and oxide particles having a first grain size distribution of an oxide selected from the group formed by the oxides of scandium, yttrium and the lanthanoids (Column 2, lines 27-29, $d_{50} > 0.9 \mu m$), and oxide particles having a second grain size distribution of an oxide selected from the group formed by the oxides of scandium, yttrium and the lanthanoids (Column 2, lines 27-29, $d_{50} \le 0.9 \mu m$).

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 3 and 4 are rejected under 35 U.S.C. 103(a) as being unpatentable over Derks et al. (US 5,075,589).

Regarding claim 3, Derks fails to disclose the limitation the oxide particles having first grain size distribution in a concentration in the range from 0.1 to 20 wt%, and the oxide particles having the second grain size distribution in a concentration in the range from 1*10⁻⁶ to 1*10⁻³

wt.%. However, Derks exemplifies the addition of oxide particles to the emissive layer in a range of 0.1-1.3wt.% (Column 2, lines 52-58), and the particles size distribution being in a range of about half the particles with $d_{50} > 0.9\mu m$ and about the other half with $d_{50} < 0.9\mu m$. Moreover, Derks recognizes the relationship and further optimization of the particles wt.% in relation to the particle size, particularly, the fact that for smaller particle sizes a smaller wt.% of those particles are required (Column 1, lines 48-52). It has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller*, 105 USPQ 233. It would have been obvious to one having ordinary skill in the art at the time the invention was made to provide the oxide particles having first grain size distribution in a concentration in the range from 0.1 to 20 wt%, and the oxide particles having the second grain size distribution in a concentration in the range from 1*10-6 to 1*10-3 wt.%, since optimization of workable ranges is considered within the skill of the art.

Regarding claim 4, Derks discloses a cathode wherein the oxide particles of the alkaline earth oxide selected from the group formed by the oxides of calcium, strontium and barium are dopes with an element selected from the group formed by scandium, yttrium and the lanthanoids in a quantity up to about 1wt.% (Column 1, lines 38-41). Derks fails to the doping elements selected from the group formed by scandium, yttrium and the lanthanoids in a quantity ranging from 0.10*10⁻⁶ to 10*10⁻⁶ wt.%. However, Derks acknowledges that for a smaller grain size it is sufficient to have smaller quantities of rare earth oxides (Column 1, lines 48-52), and furthermore, that optimum percentages by weight can be found for oxides of the rare earth metals at smaller grain sizes (Column 3, lines 34-41). It has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller*, 105 USPQ 233. It would have been obvious to one having ordinary skill in the art at the time the invention was made to provide doping

elements selected from the group formed by scandium, yttrium and the lanthanoids at a quantity ranging from 0.10*10⁻⁶ to 10*10⁻⁶ wt.%, since optimization of workable ranges is considered within the skill of the art.

Claim 7 is rejected under 35 U.S.C. 103(a) as being unpatentable over Derks et al. (US 5,075,589) in view of Saito et al. (US 4,797,593).

Regarding claim 7, Derks discloses the claimed invention except for the limitation of the electron-emitting material comprises 1 to 3 wt.% particles of an activator metal selected from the group formed by Mg, Al, Fe, Si, Ti, Hf, Zr, W, Mo, Mn and Cr. However, in the same field of endeavor, Saito discloses a cathode for electron gun further comprising an activator metal selected from the group formed by Mg, Al, Fe, Si, Ti, Hf, Zr, W, Mo, Mn and Cr (Table II) which provides partial reduction of the oxide metal and thus activates the electron-emitting layer. Thus, it would have been obvious at the time the invention was made to a person having ordinary skills in the art to incorporate the activator metal disclosed by Saito in the cathode structure of Derks in order to obtain partial reduction of the oxide metal and thus activate the electron-emitting layer.

Allowable Subject Matter

Claim 5, 6 and 8 is allowed over the prior art of record.

The following is a statement of reasons for the indication of allowable subject matter:

Regarding claim 5, the references of the Prior Art of record fails to teach or suggest the combination of the limitations as set forth in claim 5, and specifically comprising the limitation of the electron-emitting material is a stratified composite of at least a first and at least a second

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layer, said first layer comprising the oxide particles of the alkaline earth oxide selected from the group formed by the oxides of calcium, strontium and barium, and the oxide particles having the a first grain size distribution of the oxide selected from the group formed by the oxides of scandium, yttrium and the lanthanoids, and the second layer comprising the oxide particles of the alkaline earth oxide selected from the group formed by the oxides of calcium, strontium and barium, and the oxide particles having the second grain size distribution of the oxide selected from the group formed by the oxides of scandium, yttrium and the lanthanoids.

Regarding claim 6, claim 6 is allowable for the reasons given in claim 5 because of its dependency status from claim 5.

Regarding claim 8, the references of the Prior Art of record fails to teach or suggest the combination of the limitations as set forth in claim 8, and specifically comprising the limitation of the electron-emitting material comprises 1 to 3 wt.% particles of an activator metal selected from the group formed by Mg, Al, Fe, Si, Ti, Hf, W, Mo, Mn and Cr, which are coated with a metal selected from the group formed by PD, Rh, Pt, Co, Ni, Ir and Re.

Response to Arguments

Applicant's arguments filed April 15, 2004 have been fully considered but they are not persuasive.

In regards to applicant's arguments that the Derks reference fails to disclose "a coating that comprises two different grain size distributions of oxides of scandium, yttrium and the lanthanoids", the Examiner respectfully disagrees. In column 2, lines 49-52, Derks discloses "Fig. 3 shows a similar dependence of the emission process on the added quantity of yttrium oxide, which consisted of grains **half** of which had a diameter of 0.9 μ m or less (d₅₀=0.9 μ m)", accordingly, it is understood from Derks disclosure that the yttrium oxide added to the mixture

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encompassed by half of the grains and the rest of the grains exhibiting a second grain

contains two set of grain size distribution, a first a grain size distribution of 0.9 µm or les

distribution.

For the reasons given above, the rejection of claims 1-4, 7 and 9 are deemed proper.

Contact Information

Any inquiry concerning this communication or earlier communications from the examiner

should be directed to Mariceli Santiago whose telephone number is (571) 272-2464. The

examiner can normally be reached on Monday-Friday from 9:30 AM to 6:00 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Nimesh Patel, can be reached on (571) 272-2457. The fax phone number for the

organization where this application or proceeding is assigned is (703) 872-9306.

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Wハラッドリオルト Mariceli Santiago Patent Examiner Art Unit 2879

PRIMARY EXAMINER

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